EFFECT OF DIFFERENT POTASSIUM SOIL LEVELS AND FORMS OF POTASSIUM FERTILIZERS ON MICRO-ELEMETAL NUTRION STATUS OF APPLE TREES IN EARLY FRUITION PERIOD

Adam Szewczuk[1], Andrzej Komosa[2], Ewelina Gudarowska[1]

[1]Chair of Horticulture
Wrocław University of Environmental and Life Sciences
[2]Chair of Horticultural Plant Nutrition
Poznań University of Life Sciences

Abstract

The experiment was established in the spring of 1999 on grey brown podsolic soil formed from boulder clay. Apple trees of cv. Golden Delicious were planted on rootstock at 3.5×1.2 m (2381 trees·ha^{-1}). The first factor in the experiment consisted of the levels of potassium in the arable soil layer: 120, 160 and 200 mg K·kg^{-1} of soil d.m., on the basis of annual chemical analyses and determined by universal method. The second factor comprised three forms of potassium fertilizers: potassium chloride (KCl-60%), potassium sulphate (K_2SO_4) and potassium nitrate (KNO_3). Each year the analyses of macro- and microelements in the soil and leaves were made.

The increase in available potassium levels from 120 to 200 mg K·kg^{-1} of soil d.m did not have influence on the content of available forms of iron, manganese, zinc, copper and boron in the herbicide fallow strips in the tree rows or in the arable (0-20 cm) and subarable (20-40 cm)soil layers. Application of potassium sulphate fertilizer caused a significant increase in the content of zinc in herbicide strips in both soil layers compared with potassium chloride and potassium nitrate. Different potassium fertilizer forms did not cause any significant changes in amounts of available forms of iron, manganese, copper and boron.

dr hab. Adam Szewczuk, Chair of Horticulture, Wroclaw University of Environmental and Life Sciences, pl. Grunwaldzki 24A, 50-363 Wrocław, phone (071) 320 17 39, e-mail: adam.szewczuk@up.wroc.pl

The increase in available potassium levels from 120 to 200 mg $K \cdot kg^{-1}$ of soil d.m significantly reduced the content of manganese in leaves of trees. However, no significant influence on the content of iron, zinc, copper and boron in leaves of apple trees was found.

Key words: apple trees, potassium fertilizers, chlorides, sulphates, nitrates, microelements.

WPŁYW RÓŻNYCH POZIOMÓW POTASU I RODZAJU NAWOZÓW POTASOWYCH NA STAN ODŻYWIENIA JABŁONI MIKROELEMENTAMI PO WEJŚCIU DRZEW W OKRES OWOCOWANIA

Abstrakt

Doświadczenie założono wiosną 1999 r. na glebie płowej wytworzonej z glin lekkich zwałowych. Drzewa odmiany Golden Delicious na podkładce M.26 posadzono w rozstawie 3,5 x 1,2 m (2381 drzew·ha^{-1}). Pierwszym czynnikiem badań był zróżnicowany poziom zawartości potasu: 120, 160 i 200 mg $K \cdot kg^{-1}$ w ornej warstwie gleby, oznaczonego metodą uniwersalną, drugim rodzaj zastosowanego nawozu potasowego: chlorek potasu (sól potasowa 60%), siarczan potasu i saletra potasowa. Corocznie wykonywano analizy gleb i liści na zawartość mikroelementów.

Wzrost zawartości przyswajalnych form potasu w zakresie 120-200 mg $K \cdot kg^{-1}$ gleby nie miał istotnego wpływu na zawartość przyswajalnych form żelaza, manganu, cynku, miedzi i boru w pasach herbicydowych gleby, zarówno w warstwie 0-20, jak i 20-40 cm. Stosowanie siarczanu potasu istotnie zwiększało zawartość cynku w pasach herbicydowych zarówno w warstwie 0-20, jak i 20-40 cm gleby w stosunku do chlorku potasu (sól potasowa 60%) i azotanu potasu (saletra potasowa). Nie stwierdzono wpływu form nawozów potasowych na zawartość przyswajalnych form żelaza, manganu, miedzi i boru w glebie.

Wzrost zawartości przyswajalnych form potasu w glebie w zakresie 120-200 mg $K \cdot kg^{-1}$ gleby istotnie zmniejszał zawartość manganu w liściach jabłoni. Nie wykazano natomiast wpływu wzrastających poziomów potasu w glebie na zawartość w liściach jabłoni żelaza, manganu, miedzi i boru.

Słowa kluczowe: jabłoń, potas, nawozy potasowe, chlorki, siarczany, azotany, mikroelementy.

INTRODUCTION

Fertilization is the most important condition for proper growth and yielding of trees. Optimal nutrient supply for trees should take into account the vigour of the tree growth, the level of yielding as well the nutrient content in the soil and in the plant. Potassium fertilization and favourable N/K ratio can increase the frost tolerance of generative organs of fruit trees to some extent. The excess of lime can be compensated by potassium fertilization (Szűcs 2005). Potassium is taken up by apple trees in high levels, even greater than is actually needed to grow and yield properly. Excessive amounts of potassium in soil can cause worse absorption of magnesium (Lipecki and Jadczuk 1998, Pietranek and Jadczuk 2005) and calcium. Research on potassium fertilization of apple trees has brought variable re-

sults. LESZCZYŃSKI and SADOWSKI (1990) pointed out to the positive effects of K fertilization on the tree nutrient status, while PACHOLAK (1984) presented an opposite opinion.

The content of potassium and phosphorus for orchard soil is usually determined by Egner-Riehm's method and the content of magnesium by Schachtschabel's method. In horticulture, a universal method after Nowosielski is used for determining all macro- and microelements. KOMOSA and STAFECKA (2002) found out that the universal method is good for orchard soils, too. Most often, potassium is applied as potassium chloride. Being more expensive, potassium sulfate is used rarely. However, some studies conducted in Poland indicated a low sulfur level in soils of farmlands as well as in orchards (JAKUBUS 2001). That is why, using potassium sulfate for fertilization in apple orchard could by an important source of sulfur (KOMOSA, SZEWCZUK 2002, SZEWCZUK et al. 2008).

The aim of the present research has been to determine the influence of different level of potassium fertilization, as well anion accompanying potassium: chlorides, nitrites, sulfates, on the content of microelements in leaves of cv. Golden Delicious apple trees and on their nutrition status.

MATERIAL AND METHODS

The experiment was carried out in 2002-2004 years in the Experimental Station belonging to Wrocław University of Environmental and Life Sciences. In the spring of 1999, two-year-old apple trees of cv. Golden Delicious on rootstock M 26 were planted at the spacing 3.5×1.2 m (2381 trees ha^{-1}) on grey brown podsolic soil formed from boulder clay. The experiment was established in a randomized split-plot design in four replications with 4 trees per plot. The experimental plot covered 67.2 m^2 and had 16 trees, of which 4 in the middle were studied and the remaining 12 made the isolation.

Herbicide fallow strips were in the tree rows grass alleys were maintained and between them. Before planting, the macro- and microelements content of the soil was determined (Table 1). According to the content index worked out by KOMOSA and STAFECKA (2002), the following were established: N-NH$_4$+N-NO$_3$ 6-20, P 30-60, K 50-80, Ca 250-400, Mg 30.0-60 and S-SO$_4$ 10-30 mg·kg^{-1} g soil d.w. and Fe 75.0-120.0, Mn 25.0-40.0, Zn 3.0-6.0, Cu 1.0-4.0 and B 0.3-1.5 mg·kg^{-1} soil d.m. and <50 mg Cl, <50 mg Na·kg g^{-1} soil, EC <0.5 mS·cm^{-1}, i.e. low contents of nitrogen, sulfur and iron, standard amounts of phosphorus, manganese, copper and boron, high contents of potassium, magnesium, calcium, zinc, chloride and an admissible level of sodium in the arable soil layer (0-20 cm). In the subarable soil layer (20-40 cm), low contents of nitrogen, phosphorus, potassium, sulfur, zinc and copper, medium levels of calcium, magnesium, iron, manganese and boron,

Table 1

Content of macro and microelements in the soil before planting of apple trees (1998)

The soil layer (cm)	mg·kg⁻¹ soil d.w.						
	$N-NH_4$	$N-NO_3$	P	K	Ca	Mg	$S-SO_4$
0-20 20-40	3.1 t.a.*	23.4 5.5	38.0 17	96.0 22	1278 279	86.0 42	t.a. t.a.
	mg·kg⁻¹ soil d.w.						
	Fe	Zn	Mn	Cu	B	Na	Cl
0-20 20-40	66.4 72.9	9.3 2.5	25.3 39.2	2.9 0.9	0.77 0.50	6 4	55 63
	pH (H_2O)	EC mS·cm⁻¹					
0-20 20-40	6.99 6.98	0.22 0.28					

t.a.* trace amounts

and high content of chlorides, pH (in H_2O) 6.99 and 6.98, EC 0.22 (0-20 cm) and 0.18 mS·cm⁻¹ (20-40 cm) were observed.

The field trail was established as a two-factor experiment. The first factor consisted of increasing levels of potassium in the arable soil layer: 120, 160 and 200 mg K·kg⁻¹ soil d.m., based on annual chemical analyses. The second factor comprised three forms of potassium fertilizers: potassium chloride (KCl), potassium sulfate (K_2SO_4) and potassium nitrate (KNO_3). The assumed level of potassium was kept by using different rates of fertilizers (Table 2). All treatments were fertilized with nitrogen and phosphorus, according to the annual analyses of soil and leaves. Nitrogen was used as ammonium nitrate (34%) and saltpetre (13% N, 39% K-only in combination

Table 2

Rates of potassium fertilizers in 2002-2004 (mg·kg⁻¹)

Treatment	K level (mg·kg⁻¹ soil)	Fertilizer form	Sum from 1999-2001	2002	2003	2004
K-1 (KCl)	120	KCl	54	125	0	0
K-2 (KCl)	160	KCl	154	170	129	0
K-3 (KCl)	200	KCl	254	100	0	75
K-1 (K_2SO_4)	120	K_2SO_4	54	36	0	75
K-2 (K_2SO_4)	160	K_2SO_4	192	40	36	0
K-3 (K_2SO_4)	200	K_2SO_4	314	186	0	0
K-1 (KNO_3) K-2 (KNO_3) K-3 (KNO_3)	120 160 200	KNO_3 KNO_3 KNO_3	54 154 254	123 177 277	94 0 0	0 0 150

with KNO_3), phosphorus as triple superphosphate (20% P), potassium as potassium chloride (60%), potassium sulfate (41%). The fertilization was applied in the middle of March, April and May.

Soil samples from the herbicide strips of each plot were collected each year in the second half of July, separately from the layers 0-20 and 20--40 cm, by using a soil drill. Soil analyses were carried out using the universal method according to Nowosielski (1974), modified for orchard soils (Komosa and Stafecka, 2002). In this method, B was extracted in 0.03 M CH_3COOH, microelements Fe, Mn, Zn and Cu were extracted with Lindsay's solution, which contained in 1 dm $^{-3}$: 5 g $EDTAH_4$, 9.0 ml 25% ammonia, 4 g citric acid, 2 g $(CH_3COO)_2Ca \cdot 2H_2O$. Extractions were conducted in a 1:4 proportion of soil to extraction solution, (50 g dry weight soil and 200 cm^3 Lindsay's solution) for 30 minutes. After extraction, B was assessed by colorimetric analysis with curcuma, and Fe, Mn, Zn i Cu – by the AAS method (*Laboratory research method...* 1983).

Leaf samples were collected in the middle of July. One sample containing 100 leaves from the middle part of long-shoots (3-4 leaves per shoot) was collected from each plot. The concentration of Fe, Mn, Zn, Cu and B in the leaves was estimated. Mineralization of leaves for Fe, Mn, Zn and Cu assessment was performed in a mixture of HNO_3, $HClO_4$ and H_2SO_4 in the 10:1:1 ratio, while concentration of B was determined after dry digestion with calcium hydroxide (*Laboratory research method...* 1972).

The results were evaluated statistically using the analysis of variance. The significance of differences between means was evaluated according to *t*-Duncan's multiple range test at *P*=0.05.

The present study is the continuation of a previous experiment carried out in order to estimate the influence of different potassium fertilization on macro- and microelement nutrition status of young apple trees. The results, for 1991-2001 year, were published by Komosa and Szewczuk (2002). In this study, the results for the next three years are presented and concern older trees in full fruition period.

RESULTS AND DISSCUSION

The results presented in Table 3 proved that increasing potassium in arable soil layer in herbicide fallow had no influence on the content of available forms of iron, manganese, zinc, cooper and boron in soil. However, significant influence of the fertilizer forms of potassium on the content of zinc in the arable soil layer was noted (Table 3). The application of potassium sulfate caused the highest zinc content in soil, in comparison with potassium chloride and nitrate form. These differences were confirmed by the estimation of the content of available forms of microelements in the subarable

Table 3

Content of microelements in the soil layer of 0-20 cm in the herbicide strips in relation
to fertilizer form and potassium level in soil (means from 2002-2004)

K level mg·kg⁻¹ of soil	Fertilizer form			Mean
	KCl	K₂SO₄	KNO₃	
mg·Fe kg⁻¹ of soil				
120	76.7***	74.4	76.2	75.7**
160	79.4	74.4	77.6	77.1
200	68.3	69.0	56.9	64.7
Means	74.8*	72.6	70.2	
mg·Mn kg⁻¹ of soil				
120	21.1***	21.4	17.1	19.9**
160	19.2	16.8	26.5	20.8
200	25.1	19.5	11.7	18.8
Means	21.8*	19.2	18.4	
mg·Zn kg⁻¹ of soil				
120	6.2 ab	17.1 cd	8.8 abc	10.7 ab
160	6.3 ab	16.0 cd	5.3 ab	9.2 a
200	6.2 ab	16.5 cd	5.2 a	9.3 a
Means	6.2 ab	15.5 cd	6.4 ab	
mg·Cu kg⁻¹ of soil				
120	2.7***	2.8	2.6	2.7**
160	2.6	2.7	2.9	2.7
200	3.1	2.7	2.3	2.7
Means	2.8*	2.8	2.6	
mg·B kg⁻¹ of soil				
120	0.56***	0.54	0.52	0.54**
160	0.52	0.50	0.48	0.50
200	0.62	0.50	0.46	0.53
Means	0.57*	0.51	0.49	

Means marked by the same letter are not significantly different at α=0.05.
*no significant differences between the means for the fertilizer forms
**no significant differences between the means for the potassium levels
***no significant differences for the interaction fertilizer form x potassium level

Table 4

Content of microelements in the soil layer of 20-40 cm in the herbicide strips in relation
to fertilizer form and potassium level in soil (means from 2002-2004)

K level mg·kg^{-1} of soil	Fertilizer form			Mean
	KCl	K$_2$SO$_4$	KNO$_3$	
mg·Fe kg^{-1} of soil				
120	74.1***	81.4	79.0	78.2**
160	75.0	58.5	74.8	69.4
200	75.3	64.5	58.6	66.2
Means	74.8*	68.1	70.8	-
mg·Mn kg^{-1} of soil				
120	21.5***	23.4	16.7	20.5**
160	11.6	10.6	19.8	14.0
200	23.4	12.8	7.9	14.7
Means	18.8*	15.6	14.8	-
mg·Zn kg^{-1} of soil				
120	4.5ab	14.4 bc	8.0 ab	9.0ab
160	4.7ab	12.7 abc	4.4 ab	7.3a
200	4.6 ab	19.9 c	3.4 a	9.3b
Means	4.6ab	15.7c	5.3a	-
mg·Cu kg^{-1} of soil				
120	2.4***	2.5	2.3	2.4**
160	2.2	2.3	2.5	2.3
200	2.7	2.2	2.1	2.3
Means	2.4*	2.4	2.3	-
mg·B kg^{-1} of soil				
120	0.39***	0.38	0.47	0.41**
160	0.43	0.39	0.42	0.42
200	0.46	0.43	0.41	0.42
Means	0.43*	0.40	0.44	-

Means marked by the same letter are not significantly different at α=0.05.
*no significant differences between the means for the fertilizer forms
**no significant differences between the means for the potassium levels
***no significant differences for the interaction fertilizer form x potassium level

Table 5

Content of microelements in Golden Delicious apple leaves in relation to potassium level and fertilizer form in soil (means from 2002-2004)

K level mg·kg⁻¹ of soil	Fertilizer form			Mean
	KCl	K$_2$SO$_4$	KNO$_3$	
mg·Fe kg⁻¹ of soil				
120	69.1***	70.6	71.4	70.3**
160	70.3	74.7	68.1	71.0
200	72.4	69.6	74.3	72.1
Means	70.6*	71.6	71.3	-
mg·Mn kg⁻¹ of soil				
120	120.1***	112.7	131.6	121.5 b
160	111.1	112.2	104.7	109.3 ab
200	101.5	97.7	100.0	99.7 a
Means	110.9*	107.6	112.1	-
mg·Zn kg⁻¹ of soil				
120	36.0***	37.7	40.0	37.9**
160	34.8	39.9	35.4	36.7
200	35.2	47.4	35.1	39.2
Means	35.3*	41.7	36.8	-
mg·Cu kg⁻¹ of soil				
120	6.7***	6.5	6.6	6.6**
160	6.8	7.0	6.4	6.7
200	6.5	6.4	6.5	6.5
Means	6.7*	6.6	6.5	-
mg·B kg⁻¹ of soil				
120	40.9***	39.7	39.3	40.0 **
160	44.0	43.7	41.7	43.2
200	42.2	41.4	44.3	42.7
Means	42.4*	41.6	41.8	-

Means marked by the same letter are not significantly different at α=0.05.
*no significant differences between the means for the fertilizer forms
**no significant differences between the means for the potassium levels
***no significant differences for the interaction fertilizer form x potassium level

layer of soil in herbicide fallow strips (Table 4). Significantly the highest concentration of zinc was in the soil fertilized with the sulfate form of potassium fertilizers.

At the lower level of 120 and 160 mg K mg kg^{-1} of soil, the tendency to increase the content of iron, manganese and zinc in arable and subarable soil layer in herbicide fallow strips was noted. This relationship was not observed for copper and boron (Tables 3, 4).

Comparison with the content index for soils in apple orchards, worked out by KOMOSA and STAFECKA (2002), suggested that maintaining high level of potassium in soil (200 K mg·kg^{-1}), impoverished the soil of iron (change for medium into low content), manganese and zinc (no change of class content). Content of copper and boron in soil was on the medium level, regardless of the levels of potassium content in soil. Reduction of the content of iron, manganese and zinc in soil, caused by the high level of potassium, can be connected with displacing these cations from sorption complex of soil and moving them into the depth of soil profile. Reduction of content of manganese in soil by the high level of potassium (200 mg K·kg^{-1}) was confirmed by significant reduction of concentration of manganese in apple leaves.

The optimal concentration of manganese in apple leaves is 41-100 mg Mn·kg^{-1}. In the present study, the content of manganese fell from high level of content (109.3-121.5 mg·kg^{-1}) to the optimal level – 99.7 mg·kg^{-1}. No significant influence of the potassium level and potassium fertilizer forms on the nutritional status of iron, zinc, copper and boron of apple trees occurred. Similar results were obtained by GĄSTOŁ and SKRZYŃSKI (2006). According to these authors, the distribution of mineral constituents in plants depended on different organs of apple tree. It is obviously attributable to the different functions of particular fruit tree organs. GĄSTOŁ and SKRZYŃSKI [2006] did not notice the influence of dwarfing methods on leaf iron, zinc and copper content. However, different levels of microelements were noted in roots, wood and bark of apple trees (GĄSTOŁ and SKRZYŃSKI 2006). The determined content of microelements in this study was low for iron and optimal for zinc, copper and boron in comparison with the optimal contents in leaves of apple trees. The low content of iron in leaves cold be caused by high pH$_{(H_2O)}$ of soil (6.98-6.99), and thus by worse uptake of iron by plants.

CONCLUSIONS

1. The increase in the content of available potassium forms from 120 to 200 mg·kg^{-1} of soil did not have significant influence on the content of available forms of iron, manganese, zinc, copper and boron in the herbicide fallow strips or in the arable (0-20 cm) and subarable (20-40 cm) soil layer.

2. The potassium fertilizer forms affected the zinc concentration in soil. Application of potassium sulfate resulted in an increase in the zinc concentration in arable and subarable soil layers in comparison with potassium chloride and nitrate. Different potassium fertilizers forms had no influence on the iron, manganese, copper and boron concentration in soil.

3. The increase in the content of available potassium forms from 120 to 200 mg K·kg^{-1} of soil reduced significantly the concentration of manganese in leaves of apple trees. No significant interaction between increasing levels of potassium content in soil and concentration of iron, cooper, zinc and boron in apple leaves was found

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